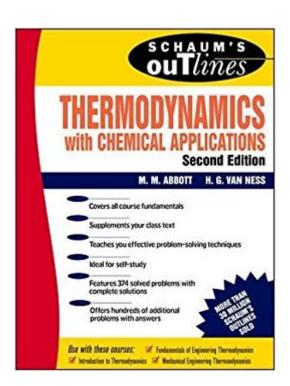


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Schaum's Outline Of Thermodynamics With Chemical Applications (Schaum's Outline Series)





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Customer Reviews

Andris A. Zoltners is professor of marketing at the J.L. Kellogg Graduate School of Management at Northwestern University. He is a founder and managing director of ZS Associates, a global business consulting firm. For over 20 years, he has served the business community as a professor, consultant, and speaker. Prabha K. Sinha is a founder and managing director of ZS Associates. He has consulted for over 200 fat and fit firms in North America, Europe, and Asia. He also teaches business executives at the Kellogg Graduate School of Management and the Management Centre Europe. Stuart J. Murphy is a visual learning specialist. He is a graduate of the Rhode Island School

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Andris A. Zoltners is professor of marketing at the J.L. Kellogg Graduate School of Management at Northwestern University. He is a founder and managing director of ZS Associates, a global business consulting firm. For over 20 years, he has served the business community as a professor, consultant, and speaker. Prabha K. Sinha is a founder and managing director of ZS Associates. He has consulted for over 200 fat and fit firms in North America, Europe, and Asia. He also teaches business executives at the Kellogg Graduate School of Management and the Management Centre Europe. Stuart J. Murphy is a visual learning specialist. He is a graduate of the Rhode Island School of Design and the Owner/President of the Management Program at Harvard's Graduate School of Business. The author of MathStart, a series of children's books that teaches mathematics, he has also coauthored a number of other leading mathematical educational programs. Rene Williams-Andriani illustrates children's books and works as a designer for Hallmark Cards in Kansas City--when she's not drawing sea lions. She received her degree in illustration from Rhode Island School of Design.

Great for reference and reviewing...

This book is great for supplemental problems. We use the Smith, Van Ness, Abbott for our course and, since this is by Van Ness and Abbott, it follows along well giving additional problems with solutions. Highly recommend all the Schaums Series for anyone needing quick reference guides.

Good seller!! Recomended.

What I expected

This is the best book you can buy, if you want to LEARN classical thermodynamics, the principles and how to apply them. Here are the highlights:(1) All basic concepts are finely presented and

explained. The discussion is complete, thorough and concise.(2) Topics covered make up a complete presentation of thermodynamics for chemical engineers, mechanical engineers and chemists. For physicists the material may be too extensive and the applications too many. A statistical-mechanical version of thermodynamics would be more appropriate for them.(3) All aspects of the theory are demonstrated with examples and applied to real engineering situations.(4) It is given much effort to teach you (a) understand fundamental thermodynamic concepts and (b) how to solve real world problems.In summary, it is a fine book, highly recommended for students, engineers and scientists mentioned in (2).

The Schaum Outline Series has stood the test of time and Thermodynamics by Abbott and Van Ness is no exception. If you want to learn thermo beyond the typical undergraduate level, progressively working through the problems chapter-by-chapter in this book will do the trick. Although this book does follow the old sign convention for work (positive when the system does work on the environment), it properly follows the specific energy convention of "energy/mass". Thus, the ideal gas equation is PV = RT, where V has units of "volume/mass". Once this simple terminology is understood, steam tables can be readily used without confusion.

Frankly, I cannot understand what the purpose of this book really is. It's certainly not a book on general or theoretical thermodynamics. If it is a textbook designed to guide the undergraduate student to understand thermodynamics, it fails. It is just not detailed enough to make the subject clear to a student. The short discussion of the theory is just that: short. Do not be mislead that you will read a few pages and understand a subject which needed centuries to be developed and still is confusing to many scientists. The problems are just not enough for an undergraduate student who wants to be good in solving problems. I had bought this book when I was a student and openned it only a few times to study. The experience was frustrating. I haven't reffered to it for years and never again will. A far better book for the undergraduate is Cengel and Boles thermodynamics.

Outline exactly what I was expecting.

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